



Dealkylation and Metallation of a Hydrocarbon in On-Surface Synthesis

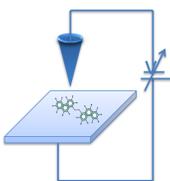
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STM @ 77 K IN UHV



Introduction: We demonstrate temperature-controlled cleavage and desorption of alkyls from carboxyl diimide picenes ($C_{12}PicDI$) on Au(111). The initially alkyl-passivated diimide group is activated and available for metal coordination.

Background

- Imide groups supports metal ion coordination^[1]
- [5]Phenacene (known as picene): polycyclic aromatic hydrocarbon

Molecular structure of $C_{12}PicDI$

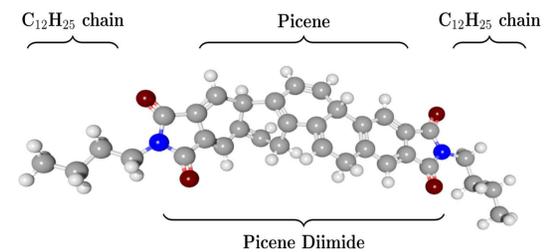
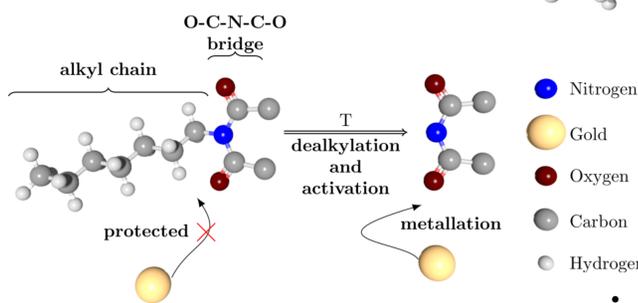
- Picene core with imide functionalities at both ends
- Dodecyl-chain-passivated imide groups

Applications

- Organic semiconductor with high electron mobility^[2a] for organic n-channel FET devices^[2b,3]

Goal

- Controlled imide activation for metal coordination

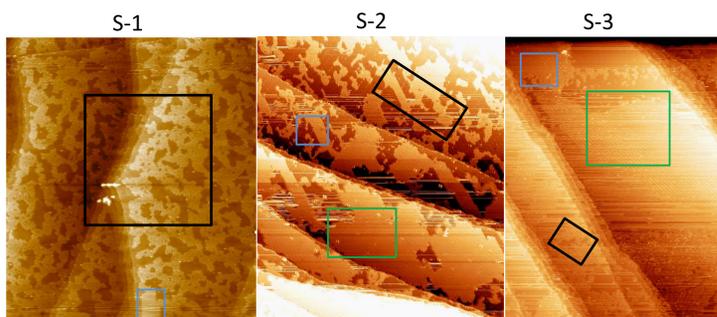


Preparations

- Three Au(111) samples prepared
- S-1: 0.45ML, surface defect density: High
- S-2: 0.50ML, surface defect density: Low
- S-3: 0.40ML, surface defect density: High

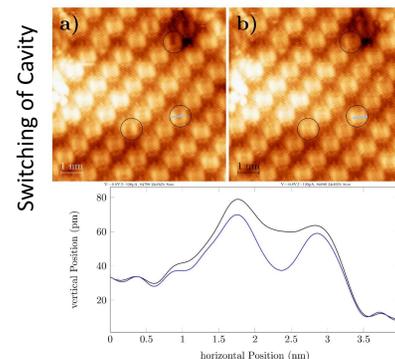
Room-Temperature preparation of $C_{12}PicDI$

*All measurements are performed at 77K



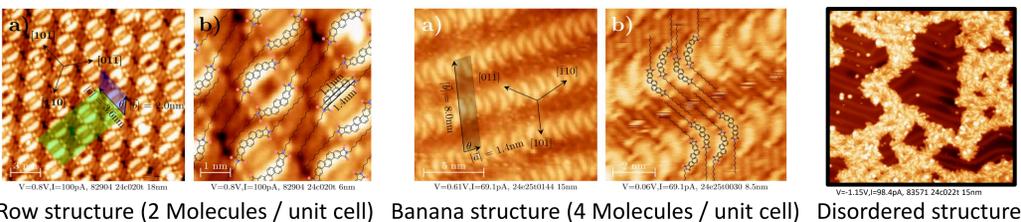
Observation:

- Molecules absorb intact on surface
- Three distinct structures (A, B, C)
- **Structure A:** Cavity-like structure
- **Structure B:** Alternatingly rotated against each other
- Structure C: Disordered
- Structure A, B: oriented in crystallographic direction => indicates strong Molecule-Substrate interaction



Switching of cavity:

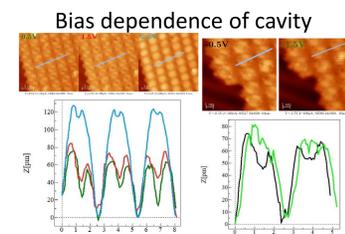
- three kinds of cavities
- white dot at centre (S1-S2-S3)
- elongated central structure (S1)
- empty (S1)
- Switching of the cavity is seen in line sections after continuous scan
- Black= bright cavity, Blue= dark cavity



Bias dependence of cavity:

Left graph +ve V/Right graph -ve V

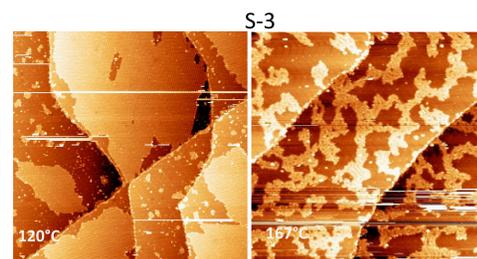
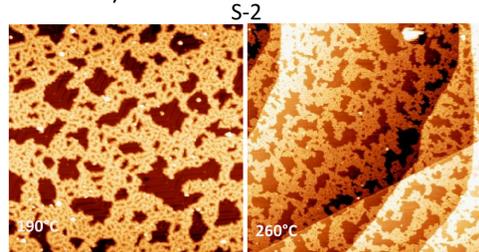
- Double peak around +0.5V, +1.5V
- Lower depth of holes at +2.5V
- Picene cores widen with increasing bias voltage



Structural Reorganization at Mild-Heating (< 320°C)

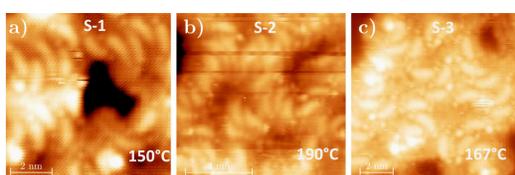
Observation (S-2):

- Only disordered structure after 190°C

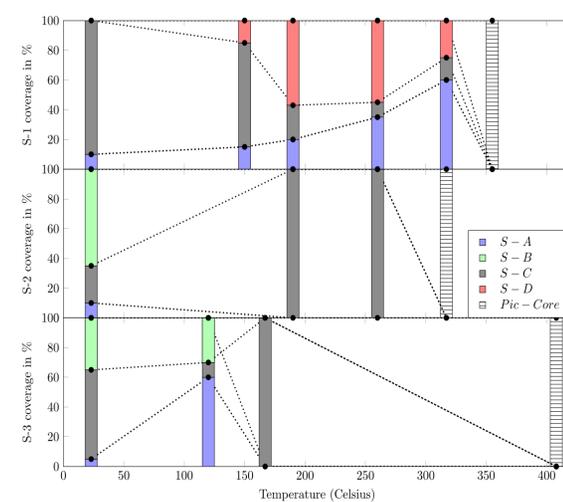


Observation (S-3):

- Ordering increases up to 120°C
- Only disordered structure after 167°C

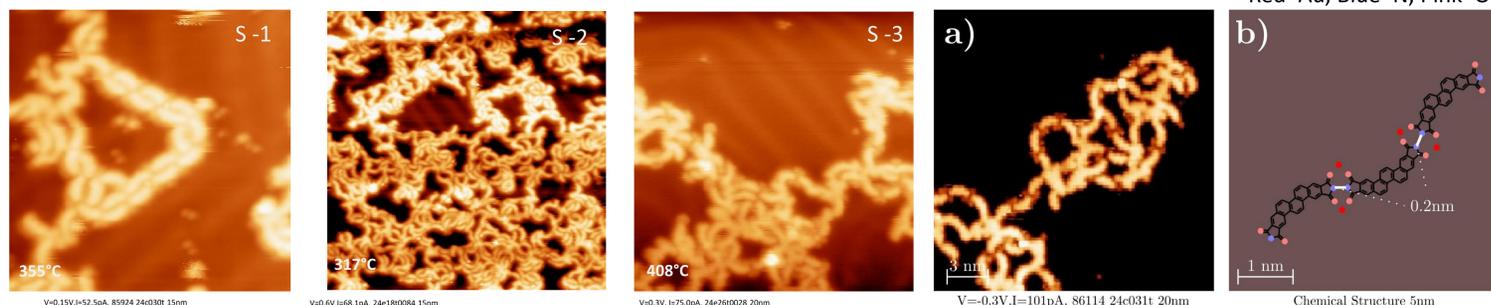
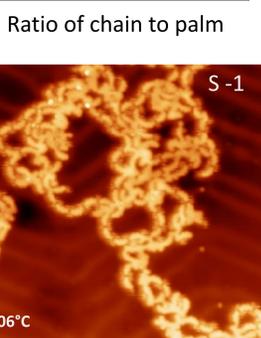
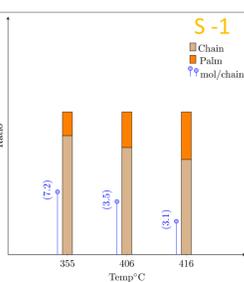


Similar diffuse structures seen for all samples



The ratio between the structures for the three samples

Dealkylation at High-Heating Treatment ($\geq 317^\circ\text{C}$)



Observations (S-1, S-2, and S-3):

- Intact diimide picenes cores without alkyl chains for all 3 samples
- Molecule-metal bonds observed as cloud-like blobs
- Alkyl chains detachment: S-1 at 355°C, S-2 at 317°C, S-3 at 408°C
- Dealkylated PicDI cores show two new structures:
 - Linear chains (lower population in S-2 and S-3)
 - 2D palm tree-shape (increases with temperature)

- N-N distance of 0.2nm for linear chains (agrees with Ref. [1])
- Four circular protrusions are seen at and between the imide head groups
- The number of molecules per chain decreases with temperature
- Alkyl chains desorb from the surface [4]

Interpretation:

- Our results align with previous work^[1] regarding molecule-molecule distance and circular protrusion shape
- Au atoms are extracted and are involved in the chain formation process

Summary

Reference

- Sample variations may stem from deposition temperature differences, cleanliness, and heating rates due to filament change after initial preparation
- Mild heating induces structural changes, with intact molecules still present on the surface
- Diimide picene cores remain intact even after high-temperature treatment
- We've shown that the alkyl chain can be removed without damaging the picene diimide core, potentially enabling the creation of high-performance n-channel organic semiconductors through on-surface synthesis

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[1] Yu, M., Xu, W., Kalashnyk, N. et al. From zero to two dimensions: supramolecular nanostructures formed from perylene-3,4,9,10-tetracarboxylic diimide (PTCDI) and Ni on the Au(111) surface through the interplay between hydrogen-bonding and electrostatic metal-organic interactions. *Nano Res.* 5, 903–916 (2012)

[2a] Ruiz, R.; et al. Pentacene thin film growth. *Chem. Mater.* 2004, 16, 4497–4508; [2b] Nakayama, Y. et al. Epitaxial growth of an organic p-n heterojunction: C_{60} on singlecrystal pentacene. *ACS Appl. Mater. Interfaces* 2016, 8, 13499–13505.

[3] Guo, Y.; et al. Facile synthesis of picenes incorporating imide moieties at both edges of the molecule and their application to n-channel field-effect transistors. *RSC Advances.* 2020, 52, 31547–31552

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